

Photoreactions of $W(CO)_5$ pyridine in an Ar Matrix at 10 K. Evidence for the Formation of C_5 - $W(CO)_4$ -pyridine after Short Wavelength Photolysis

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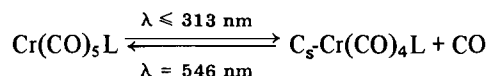
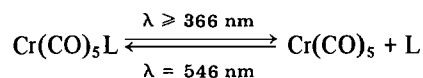
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Photolysis of $W(CO)_5$ pyridine in an Ar matrix at 12 K with $320 \text{ nm} < \lambda < 390 \text{ nm}$ resulted in the rupture of the ligand, indicating that bulky ligands can be generated in matrices [1, 2]. The same was found for $W(CO)_5PCl_3$ [3], independent of the wavelength of irradiation. Photolysis of $Cr(CO)_5L$ complexes (L = pyridine, pyrazine or NMe_3) however was shown to be dependent on the wavelength of irradiation and two complexes were observed for the first time in matrices [4, 5]:



This photochemical behaviour was explained by a one electron d-orbital energy diagram. Long wavelength photolysis results in occupation of the d_{z^2} σ antibonding orbital and formation of $Cr(CO)_5$ and free ligand (L) is the result. Short wavelength photolysis results in occupation of the σ^* $d_{x^2-y^2}$ orbital and labilization along the x and y axis, yielding C_5 - $Cr(CO)_4L$ and free CO. In contrast photolysis of $Cr(CO)_5PCl_3$ appeared to be wavelength independent. From the combination of UV-visible and PES data it was deduced that in the case of the PCl_3 complex, the energy difference between the $d_{x^2-y^2}$ and d_{z^2} orbitals is low. This small energy difference makes a fast non-radiative decay from the $d_{x^2-y^2}$ to the d_{z^2} orbital possible and only rupture of the ligand PCl_3 was found. From $W(CO)_5$ pyridine, even irradiated between 320 and 390 nm, no traces of other photo-products were detected. In this study we describe the short wavelength photolysis of $W(CO)_5$ pyridine ($\lambda = 229$ and 254 nm) and we find evidence for the formation of C_5 - $W(CO)_4$ pyridine.

Short wavelength photolysis ($\lambda = 229$ and 254 nm) resulted in the formation of free CO and C_5 - $W(CO)_4$ -pyridine, deduced from the four new CO stretching bands at 2033.1 , 1912.9 , 1906.2 and 1879.6 cm^{-1} ($3 A' + A''$) (see Fig. 1 and Table I).

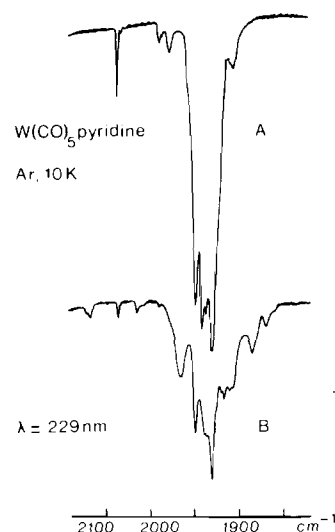


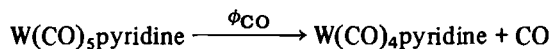
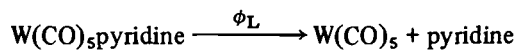
Fig. 1. Infrared spectrum of $W(CO)_5$ pyridine in an Ar matrix at 10 K. A) After deposition. B) After 100 min photolysis with $\lambda = 229 \text{ nm}$.

TABLE. Infrared Frequencies in cm^{-1}

Assignments	A_1	B_1	$E + A_1$	$E(^{13}CO)$
C_{4v} - $W(CO)_5$ pyridine Ar matrix, 10 K	2076.8	1974.7	1947.2 1939.9 1933.9 1926.9	1902.6
Assignments	A_1	E	A_1	
C_{4v} - $W(CO)_5$, Ar matrix, 10 K	2097.3	1962.4	not observed	
Assignments	A'	A'	A'	A''
C_5 - $W(CO)_4$ pyridine Ar matrix, 10 K	2033.1	1912.9	1906.2	1879.6

These bands closely correspond to the bands measured for C_5 - $Cr(CO)_4$ pyridine: 2032.2 , 1915.9 , 1907.8 and 1881.5 cm^{-1} (split into 1884.6 and 1878.3 cm^{-1}). In contrast to the results with the $Cr(CO)_5$ pyridine complex, almost the same quantum yield was found for the formation of $W(CO)_5$ (2097.3 and 1962.8 cm^{-1}) after short wavelength photolysis.

The third band (A_1) of $W(CO)_5$ at about 1931 cm^{-1} was obscured by parent bands. Solution photochemistry has already shown that the quantum yield for loss of CO occurred with a very low efficiency, even after irradiation at $\lambda = 254\text{ nm}$ [6]:



$$\phi_L = 0.63 \text{ at } \lambda = 436\text{ nm} \text{ and } 0.34 \text{ at } \lambda = 254\text{ nm}.$$

$$\phi_{CO} = 0.002 \text{ at } \lambda = 436\text{ nm} \text{ and } 0.04 \text{ at } \lambda = 254\text{ nm} \text{ in iso-octane-pent-1-ene (2:1 v/v)}.$$

The high quantum yield of $W(CO)_5$ formation can be explained by increased non-radiative decay from the $d_{x^2-y^2}$ to the excited d_{z^2} orbital in going from $Cr(CO)_5\text{pyridine}$ to $W(CO)_5\text{pyridine}$. The result is an increased loss of the ligand pyridine. This increased non-radiative decay may be caused by the increased spin-orbit coupling for the $W(CO)_5\text{pyridine}$ complex. The same was found for the photolysis in Xe matrices; externally enhanced spin-orbit coupling resulted in low quantum yields and fast non-radiative decays [5, 7]. This is illustrated by the phosphorescence lifetime (τ_p) for C_6H_6 in matrices at 10 K. This lifetime is very much shorter for Xe, than for Ar (0.02 versus 14.4 sec) and fluorescence is entirely

quenched in Xe, whereas the fluorescence in Ar can be measured (46 msec) [7]. It should also be noticed that in the series $M(CO)_5\text{pyridine}$ ($M = Cr, Mo, W$) the Mo and W complexes show luminescence after irradiation into the $\approx 400\text{ nm}$ bands (d-d transition), whereas the Cr complex does not luminesce [8].

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